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atoms of hydrogen-1 are in a
1.007-g sample? Recall that
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1.007 amu is the mass of one atom of hydrogen-1. Round your answer to two significant digits. 1.007 g 1 H-1 atom \_\_ 1.672 10 24 g a. 6.023 1023 H-1 atoms 4.

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sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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react to yield hydrogen bromide.

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